

Topic 4: THE CELL MEMBRANE

I Membrane Structure and Composition

A Lipid Bilayer

- 1 Phospholipids are primary component
 - a fatty acid chains (hydrophobic) point into the membrane
 - b glycerol, phosphoric acid (hydrophilic) point outside the membrane (= cell interior or cell exterior)
- 2 Cholesterol is embedded
 - a more cholesterol = more fluid membrane, but better barrier to H₂O
- 3 Physical barrier, also hydrophobic interior is difficult to pass through
- 4 Fluidity – a medium (solvent) for other substances to function in

B Proteins

- 1 span membrane: hydrophilic regions at ends, hydrophobic central region.
- 2 stud outside or inside – may interact with other membrane components by ionic interactions or hydrogen bonds
- 3 Channels for small water soluble molecules, e.g., ions
- 4 Receptor sites for hormones, neurotransmitters, etc. (ligands = molecules that bind receptors)
- 5 Membrane bound enzymes
- 6 Link membrane to cytoskeleton
- 7 Cell adhesion molecules
- 8 Recognition markers

C Carbohydrates

- 1 only on outer surface
- 2 glycoproteins
- 3 glycolipids
- 4 modify/change function of proteins they are linked to
- 5 recognition markers
- 6 cell adhesion

D Fluid Mosaic Model

II Cell to Cell Adhesions

A Extracellular matrix

- 1 meshwork of fibery proteins

- 2 acts as glue, and to maintain proper spacing between cells

B Cell junctions

- 1 desmosomes
 - a place where filaments tie cells together
 - b can form continuous protein network of intracellular proteins with extracellular proteins
- 2 tight junctions = adherens junction
 - a impermeable junctions that join lateral edges of epithelial cells
 - b allows epithelial cells to have “sides”, and to themselves act as specialized selectively permeable barriers
- 3 gap junctions
 - a gaps between cells joined by tunnel composed of protein

C Other cell adhesion molecules (CAMs)

- 1 cell-matrix adhesion
- 2 cell-cell adhesion (e.g. N-CAM)
- 3 some are loose: allow rapid binding/unbinding for movement
- 4 some are tight and permanent

III Membrane Transport

A Overview

- 1 Plasma membrane is selectively permeable
- 2 Factors that affect **permeability** of a molecule across LIPID BILAYER
 - a solubility in lipid
 - i gases (O₂, CO₂), fatty acids cross
 - ii ions (Na⁺) and hydrophilic molecules don't
 - iii H₂O?
 - b size: smaller easier to cross
- 3 Presence of membrane proteins allows additional (selective) permeability because of
 - a channels
 - b carriers

B Passive Diffusion

- 1 Requires no energy
- 2 Substances move down/with concentration gradient

- 3 Factors that affect **rate of diffusion**
 - a magnitude of **concentration gradient** – greater the difference in concentration, the greater the rate of diffusion
 - b permeability of membrane to substance
 - c surface area of membrane – greater surface area = greater rate of diffusion
 - d molecular weight of substance – small particles have a higher rate of diffusion
 - e distance over which it occurs

4 By diffusion across bilayer (gases, lipids)

5 By diffusion through channels (ions)

6 By osmosis

- a “Special case” of diffusion
- b Net diffusion of water down its concentration gradient
- c Or “up” the concentration gradient of all other solutes combined = up the osmolarity gradient
- d Water moves across lipid bilayer OR through channels

3 By carrier Mediated Transport = “facilitated diffusion”

- a Carrier proteins span membrane
- b No energy required
- c Assists movement of substance down concentration gradient

C Active Transport

1 Requires energy (ATP)

2 Can move molecules against/up concentration gradient

3 By “primary active transport”

- a called pumps (also called ATPases)
- b Example: Na⁺/K⁺ ATPase pump: pumps Na⁺ out; K⁺ in

4 By secondary active transport

- a An energy-requiring process is linked to one that does not require energy
- b frequently coupled to the Na⁺/K⁺ ATPase
- c Na⁺/K⁺ ATPase makes K⁺ high inside, Na⁺ high outside
- d additional carrier molecules couple transport with the re-intake of Na⁺

- 5 Special case of secondary active transport: transport across epithelia
- a epithelia with adherens junctions have 'sides' with different distributions of carrier proteins, pumps, and channels
 - b allows selective transport of materials
 - c example: glucose from intestinal lumen to blood

Vesicular transport

- 1 Transport in or out of cell in membrane bound vesicle
- 2 Endocytosis: surrounding & engulfing of substance
 - a two fates of endocytosed material:
 - i endosome fuses with lysosome and material is digested/released within the cell
 - ii endosome travels across the cell to be released on the other side, can be used to transport material across a cell layer
- 3 Exocytosis: fusion of vesicle with membrane & release
 - a mechanism of secretion
 - b mechanism of maintaining the plasma membrane

Topic 5: MEMBRANE POTENTIAL; REGULATION/COORDINATION OF CELLULAR ACTIVITIES

I Membrane Potential

A Overview

- 1 Definition: a separation of charges across a membrane (unequal distribution)
- 2 Energy is required to separate charges that want to be together (opposites attract)
- 3 This means that separated charges have potential to do work
- 4 Unit of measure is mV (1/1000 of a volt)

B All living cells have membrane potential

- 1 Slight excess of positive outside
- 2 Slight excess of negative inside
- 3 Ions responsible: Na^+ , K^+ , A^- (A^- is large intracellular particles = anions)

C Causes of Membrane Potential in Living Cells

- 1 Effect of K^+ alone (hypothetical situation: cell filled with only K^+ and A^- , in a bath of Na^+ and Cl^- , membrane permeable to only K^+)
 - a Concentration gradient stronger than Electrical gradient
 - b So K^+ diffuses out of cell
 - c A^- stays in cell
 - d Results in increase of electrical gradient
 - e Continues until electrical gradient balances concentration gradient:
 - f balance is called equilibrium potential = -90mV
 - g this is a negative potential because of the net negative charge inside the cell
- 2 Effect of Na^+ alone (hypothetical situation; same cell as above, but membrane is permeable to only Na^+)
 - a concentration gradient stronger than electrical gradient
 - b Na^+ diffuses into cell
 - c Cl^- stays outside
 - d diffusion of Na^+ continues till a balance is achieved
 - e equilibrium potential = $+60\text{mV}$
 - f this is a positive potential because of the net positive charge inside the cell

- 3 All together now; “real cell”
 - a Membrane is more permeable to K^+ than Na^+
 - b K^+ wants to move out of cell with its own concentration grad but against electrical gradient.
 - c Na^+ wants to move in with its own concentration gradient and its own electrical gradient
 - d membrane is more permeable to K^+ so more K^+ moves out relative to Na^+
 - e but enough Na^+ moves in to reduce membrane potential from -90 to $-70mV$
 - f this is still hypothetical – something needs to keep Na^+ high outside and K^+ high inside
 - g Na^+K^+ ATPase pump counterbalances diffusion of K^+ and Na^+ to maintain the -70 potential
 - h ALL PASSIVE FORCES BALANCE BY ACTIVE FORCES; NO NET DIFFUSION

- 5 Cl^-
 - a Does not influence membrane potential
 - b Is influenced by membrane potential
 - c Cl^- diffuses passively across membrane until its concentration gradient is balanced by overall electrical gradient established by membrane potential

II Cell-Cell Communication

- A Types of communication
 - 1 direct communication
 - a gap junctions: small molecules pass through “tunnel” between cells
 - b cell-surface communication, direct contact of cell-surface proteins

 - 2 diffusible signaling molecules
 - a paracrine
 - i short-range diffusion
 - ii neurotransmitters = specialized type of paracrine signaling
 - b endocrine
 - iv long-range diffusion via blood

B Action of chemical messengers (ligands) on cell-surface receptors

- 1 initial effect on receptor is a conformational change
- 2 direct effect on receptor
 - a receptor is a channel (e.g. Ach receptor = Na⁺ channel)
 - b receptor is an enzyme (e.g. phosphatase, receptor then adds phosphates to other proteins to change their activity)
- 3 second messenger systems
 - a activates a cascade of chemical reactions
 - b cyclic AMP and Ca⁺⁺ are common second messengers

III Regulation of cellular activities. Consider an example of a secretory cell regulating production of a hormone.

A Homeostatic maintenance of a desired level of hormone

- 1 End-product inhibition: end product of metabolism (hormone) is an inhibitor of an enzyme in the metabolic pathway
 - a metabolic pathway: series of enzyme-catalyzed steps that result in a final product
- 2 Autocrine signaling: end product of metabolism (hormone) acts on a cellular receptor – binding leads to inhibition of an enzyme in the pathway
- 3 Both are examples of negative feedback at the cellular level

B Regulated change of hormone levels

- 1 Activity of enzyme is changed (inhibited/stimulated)
- 2 Amount of enzyme is changed
 - a change in rate of protein synthesis (RNA → protein is inhibited)
 - b change in rate of gene transcription (change in gene expression; DNA → RNA is inhibited)
- 3 Other processes can be regulated
4. How is the change accomplished?
 - a. An additional signal from outside the cell binds to a receptor, initiates change
 - b. A number of mechanisms within the cell can also result in change